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Boiling Points

*Intermolecular forces and
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Dispersion Interactions
Partial Pressures \u0026amp;

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SECTION 1 Date CHAPTER 11
REVIEW Gases Class SHORT
ANSWER Answer the following
questions in the space
provided. b Pressure – orce
For a constant force, when
the surface area is tripled

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Answers
the surface area pressure is
(a) doubled. as much. (c)
ripled. 7-0 (d) unchanged.

Rank the following pressures
in increasing order. (c) 76
torr (a) 50 kPa 0, 00ictbv-x

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Schools *Answers*

If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B.

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Answers
gases are more compressible.
C. the particles move faster
in a gas and there is a
greater distance between
them.

*Chapter 11 Gases Review
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Answers Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This

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Answers describes how mixing gases affects the properties of the resulting mixture.

Chapter 11 - Gases

462 Chapter 11 Gases

Discovering the

Relationships Between

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Answers If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume,

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Answers
temperature, or number of gas particles) inside and outside the balloon.

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CHAPTER 11 REVIEW Gases
SECTION 1 SHORT ANSWER

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Answer the following questions in the space provided. 1. b Pressure surface area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c)

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Answers (d) unchanged. 2.
d, c, a, b Rank the
following pressures in
increasing order. (a) 50 kPa
(c) 76 torr (b) 2 atm (d)
100 N/m² 3.

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Gases: Section 1: Gases and
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*Chapter 11- Gases: Section
1: Gases and Pressure ...*

this theory explains some of the properties of ideal gases. In this chapter, you will study the predictions of kinetic-molecular theory for gases in more detail.

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This includes the relationship among the temperature, pressure, volume, and amount of gas in a sample. SECTION 11.1 Key Terms pressure newton barometer millimeters of mercury

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CHAPTER 11 REVIEW Gases
SECTION 2 SHORT ANSWER

Answer the following
questions in the space

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Answers provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same

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Answers
decrease b. volume
increases, temperature stays
the same

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CHAPTER 11 REVIEW Gases

SECTION 1 SHORT ANSWER

Answer the following
questions in the space

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Answers. 1. _____ Pressure = . For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. _____ Rank the following pressures in

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