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2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas. $2AlBr_3 + 3Cl_2 \rightarrow 2AlCl_3 + 3Br_2$

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to react with five liters of nitrogen to produce ammonia? (Assume constant temperature and pressure.) 5 L 3 "lids l- l 2. What volume of ammonia is produced in the reaction in Problem 1?

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WHS AP Chemistry 5 The Gas Laws

CHARLES'S LAW Charles' Law states the volume of a gas varies directly with the Kelvin temperature, assuming the pressure is constant. And a variation of the law states that pressure of a gas varies directly with the Kelvin

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