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- Part-1 - Chemistry **Changes**

**of States of Matter Chemical
and Physical Changes**

States of Matter : Solid

Liquid Gas *10 Amazing*

Experiments with Water

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Pure Substances and **Chapter 13 Answer**

Mixtures! (Classification of
Matter) Physical and Chemical

Changes

Understanding Atoms,
elements, and molecules Part
#1 (9min)

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Physical and Chemical

Changes: Chemistry for Kids

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\u0026 Changing State #20

Pure Substances and

Mixtures, Elements \u0026

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atmosphere, ozone formation,

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1.2 Chemistry and Matter

Chemistry Central Science

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Chapter 1 Section 2 MATTER

and Its Properties Matter -

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Anything that has MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element. ELEMENT- a pure substance

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Chemistry Matter And

made up of only one type of
atom.

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Section 1

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• Chapter 3 35 Section 3.1

Properties of Matter pages

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70–75 Problem-Solving Lab 1.
Explain why the flow of a compressed gas must be controlled for practical and safe use.

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eight 3. nine 4. accepted 5.

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Electrons in Atoms Reviewing
Vocabulary Match the
definition in Column A with
the term in Column B. Column
A Column B 1. The set of
frequencies of the

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electromagnetic waves
emitted by the atoms of an

element 2. The minimum

amount of energy that can be

lost or gained by ...

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Properties of Matter - 2.1

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Chemistry: Chapter 2: Matter
and Change. extensive
property. intensive
property. extensive
properties (examples)
intensive property

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(examples) a property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not... mass, volume, length, weight.

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number of moles of known

substance C H C H C H O O O

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of moles of unknown

substance O CO H O C H CO H

O C H O H O C H O CO use

mole ratio 5 mol O 2 1 mol C

3 H 8 3 mol CO 2 1 mol C 3 H

8 4 mol H 2 O 1 mol C 3 H 8

1 mol C 3 H 8 5 mol O 2

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MATH SKILLS TRANSPARENCY

MASTER 16

A “chemical change” is a change that produces matter with a different composition than the original matter.

Chemical Change A change in

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Key
which one or more substances
are converted into different
substances. Heat and light
are often evidence of a
chemical change. Properties
of Compounds Quite different
properties than their
component elements.

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