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~~Base and Salt ||~~

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pH scale is a

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measure of how strong or weak an acid or alkali is. It is a continuous scale from below 0 to above 14. The colour chart below shows the colour that

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an acid or
alkali is. It is
a continuous
scale from below
0 to above 14.
The colour chart
below shows the

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colour that
universal
indicator turns
when it's mixed
with solutions
of different
pHs: Neutral
solutions such
as water have a
pH of 7. Acids
have a pH of
less than 7.

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Understand the
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B. Find the pH
of a 0.00476 M
hydrochloric
acid solution.

2. A. Write the
equation for the
dissociation of
sulfuric acid.

B. Find the pH
of a solution

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that contains
3.25 g of H_2SO_4 dissolved in
2.75 liters of
solution. 3. A.

Write the
equation for the
dissociation of
sodium
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Find the pH of a
0.000841 M
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Name: What is

pH? pH is a

measurement of

how acidic or

how basic a

solution is. The

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pH scale starts at 0 and goes up to 14. Halfway between 0 and 14 is 7, which is neutral.

Compounds are acidic if they have a pH lower than 7.

Compounds with a pH higher than 7 are said to be basic or

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alkaline.
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of Matter 1.

Name the
physical states
in which almost
all matter
exists. 2. In
which state(s)
of matter are
the molecules
most compressed?

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In which state(s) of matter do the molecules fill the entire volume of a container? 4.

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items to
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acids and bases.

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excess reagent.

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