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electrons does a single neutral
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Considering your answer to question 5, how many electrons can fit in the entire 5th energy level? can (A_{Vel} is a ha AC5a) 5th 7. Write the notation for an electron spinning clockwise in a p

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elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

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CF bond is polar covalent. (b)

$|\Delta EN| = |EN_{Cl} - EN_C| = 3.0 - 2.1$

$|\Delta EN| = 0.9$. Since $0.5 < EN_{Cl} - EN_C$

< 1.7 , the PCl bond is polar

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covalent. (c) !ENCl"Li Section 8.5
Electronegativity and Polarity 265
Compare and contrast polar . that
is in excess.

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an s orbital and the principal

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energy level in which it is found?
3. What is the shape of a p orbital? How many p orbitals are there in a sublevel? 4.

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